Ion Gauge Gas Correction Factors

The sensitivity of an ionization gauge is related to the ionization rate of the gas being measured. Small atoms are more dificult to ionize, thus give lower readings, and vice versa for larger, heavier atoms.

| Gas | <u>Symbol</u> | Factor |
|---------------------|-------------------------------|--------|
| Helium | He | 0.18 |
| Neon | Ne | 0.30 |
| Deuterium | D ₂ | 0.35 |
| Hydrogen | H ₂ | 0.46 |
| Nitrogen | N ₂ | 1.00 |
| Air | | 1.00 |
| Oxygen | 02 | 1.01 |
| Carbon Monoxide | CO | 1.05 |
| Water | H ₂ O | 1.12 |
| Nitrogen Oxide | NO | 1.15 |
| Ammonia | NH ₃ | 1.23 |
| Argon | Ar | 1.29 |
| Methane | CH ₄ | 1.40 |
| Carbon Dioxide | CO ₂ | 1.42 |
| Krypton | Kr | 1.94 |
| Sulfur Hexaflouride | SF ₆ | 2.20 |
| Ethane | C ₂ H ₆ | 2.60 |
| Xenon | Xe | 2.87 |
| Mercury | Hg | 3.64 |

Example: In a gauge calibrated for nitrogen, where the dominant gas is other than nitrogen, divide the reading by the appropriate gas ionization factor.